

## Automated synthesis of [ $^{18}\text{F}$ ] FBEM for labeling of thiol containing compounds

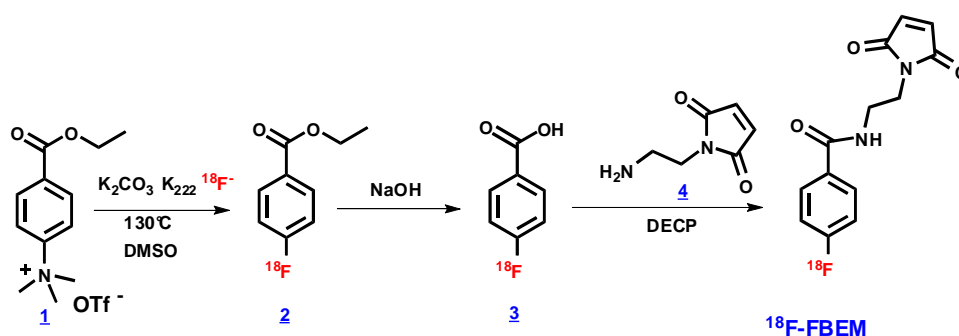
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**Objectives :** [ $^{18}\text{F}$ ]FBEM, i.e. *N*-[2-(4-[ $^{18}\text{F}$ ]fluorobenzamido)ethyl]maleimide, is a useful synthon employed for the specific radiolabeling of thiol containing compounds, including peptides and proteins <sup>[1]</sup>. The aim of the present work was to develop a fast, reproducible and fully automated synthesis of this compound in order to improve its availability as well as for obvious radioprotection matters.

**Methods :** A three-step synthetic pathway was followed (scheme 1) and implemented on a GE FastLab<sup>TM</sup> system by modifying the original [ $^{18}\text{F}$ ]FDG sequence and reagents cassette configuration. The process starts with the [ $^{18}\text{F}$ ]  $\text{F}^-$  nucleophilic substitution of the trimethylammonium ethylbenzoate compound 1 followed by NaOH hydrolysis performed in the same labeling reactor <sup>[2]</sup>. After acidification (HCl 0.25M), the resulting [ $^{18}\text{F}$ ]Fluorobenzoic acid 3 was trapped and purified on a solid phase extraction cartridge before being coupled to amino-maleimide compound 4 in the next step. This was carried out using diethylcyanophosphonate <sup>[3]</sup> in acetonitrile at 70°C. Then [ $^{18}\text{F}$ ]FBEM could be isolated and purified on a second SPE cartridge.



**Scheme 1 :** Radiosynthesis of [ $^{18}\text{F}$ ]FBEM

**Results :** The fully automated process takes around 55 minutes and the desired product is obtained with a decay-corrected radiochemical yield of 41 % (n=11) and a radiochemical purity  $\geq 90$  % as determined by HPLC and TLC. Subsequent conjugation to thiol containing compounds was also carried out.

**Conclusions :** A completely automated radiosynthesis of [ $^{18}\text{F}$ ]FBEM has been developed with good radiochemical yields and purity. The resulting SPE purified maleimide synthon is suitable for the labeling of various thiol containing compounds under mild conditions.

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**References :** [1] D.O. Kiesewetter, et al., (2008) *J. Fluorine Chem.*129, 799-806. [2] D.Thonon, et al., (2011) *Mol. Imaging Biol.*, DOI : 10.1007/s11307-011-0470-x. [3] .O. Kiesewetter, et al., (2011) *Applied Rad. Isotopes* 69, 410-414